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(54) Compositions for use in cleaning teeth containing at least one of yttrium scandium or lanthanum and the lanthanides

(57) Teeth are cleaned by applying thereto compositions, such as mouthwashes, toothpastes and dental gels, containing one or more cations of elements selected from yttrium, scandium and lanthanum and the lanthanides.

SPECIFICATION

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Method of cleaning teeth and compositions for use in such method

5	This invention relates to dental hygiene, and more particularly it relates to a method of cleaning	5
10	teeth. It is known that a wide variety of elemental cations, including those derived from gallium, this known, lead, indium, beryllium, cerium, dysprosium, iron, arsenic, thorium, iridium, rubidium, vanadium, zironium, tian, allumi, aluminium, copper, ruthenium, gold, samarium, rhodium, mercury, lithium, cobalt, lanthanum, uranium, zinc, barium, castium, stontium, calcium, cadmium and chromium, reduce the acid solubility of tooth enamel in an in vitro test system (R.S. Manly and B.G. Bibby, J. Dent. Res., 1949, 28, 180–171). It is also known that tooth enamel may be remineralised by sequential application to the testh of a cationic and an	10
15	anionic component which react below the tooth surface to form an insoluble saft. Examples or the cationic component are lons derived from barium, lanthanum, manganese, lead, tin, zinc, indium, zirconium, iron, titanium, vanadium and cadmium (UK Patent Specification No. 1465.1726).	15
20	It has been found that administration of yttrium nitrate to rats, either by intrapertroneal injection or via the drinking water, reduces the incidence of caries and this effect has been attributed to the incorporation of yttrium into the dental enamel, thus reducing its acid solubility. (R. Castillo Mercado and T.G. Ludwig, Archs.oral Biol., 1973, 18, 637–640), It has also been found that mouthrinsing with a stanous fluoride solution reduces plaque formation on tooth enamel (N. Tinanoff, J.M. Brady, and A. Gross, Caries Res., 1978, 10, 415–426). In Tinanoff,	20
25	J.Dent.Res., 1977, 56, Spec. Issue A, A138). It has now been discovered, and herein lies our invention, that deposits such as dental plaque may be removed from the surface of teeth, or may be prevented from adhering thereto, by	25
30	According to the invention there is provided a method of cleaning teeth by applying merero cation of one or more elements selected from yttrium, scandium and lanthanum and the lanthanides. The fanthanides include cerium, praseodymium, neodymium, promethium, samarium, europium, displaim, gadolinium, terbium, thugin, estibum, thujium, ytterbium and lutetium. These elements are very closely related chemically since they all fall with Group IIIb of the Periodic Table. Lanthanum and the lanthanides form the first group of Inner Transition.	30
35	Elements within Group IIIb. The method of the invention is particularly useful for cleaning teeth in human beings. The efficent cleaning of teeth is, of course, of immense cosnetic value. One of the substances routinely found on the surface of teeth is bacterial plaque and the method of this invention is particularly useful in removing plaque from teeth or in preventing its adhering thereto. The	35
40	method of the invention is also useful in removing various types of stain from teem, for example the stain produced by smoking tobacco. In addition, bacterial plaque is generally regarded as e dominant etiological factor in caries and periodontal disease and removal of plaque from teeth or prevention of its accumulation is known to have a beneficial effect in those conditions.	40
45	reduction in the incidence of caries and/or periodontal disease. The preferred cations for use in the method of the invention are those derived from lanthanum and the larthanides and a particularly preferred cation is the lanthanum cation.	45
50	When used in the method of the invention, the cation may be in the form or a sait, and preferably in the form of a water-soluble salt, Examples of such water soluble salts are the chloride, bromide, iodide, nitrate, acetate or sulphate. The cation may also be used in the form of a cett with an arithmetrial anion.	50
55	The preferred compound for use in the method of the invention is lanthanum chloride, LaCl ₃ . The amount of cation used in the method of the invention may vary from 0.01 m.moles to 1 m.moles of cation and preferably from 0.1 m.moles to 0.5 m.moles of cation, and it may be applied from once a week to 1 to 10 times per day. A preferred regime is three times per day, after meals, or failing this, twice per day, night and moming.	55
60	arter meals, or raing ints, twice per locy, lingth an informing result simply by application of the cation. The method off the invention achieves a satisfactory result simply by application of the cation to the teeth, for example in the form of a simple aqueous solution. However, an improved degree of cleaning can be achieved if the method of the invention is combined with one or more of the normal mechanical methods of cleaning teeth, for example if combined with the use of a toothbrush, toothpick, dental floss, dental probe or rotary dental brush. A particularly preferred adjunct to the method of the invention is the use of a toothbrush.	60
65	preferred adjunct to the method of the invention is the use of a bodinate. The cation for use in the method of the invention may be presented in the form of a composition such as a simple aqueous solution or suspension or in the form of a more	65

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-	sophisticated composition such as a mouthwash, toothpaste, prophylaxis paste, toothpowder, pastille, chewing gum or oral spray, or it may be incorporated into a beverage, nutritional substance or confection, it may also be incorporated into the public water supply. The compositions described above are those which are well known to those skilled in this art. They may incorporate any of the ingredients normally used in such compositions, with the	5
5	Iney may incorporate any or me ingredents normally used an such compositions, with ine addition of the cation in the form of a salt. In the case of a mouthwash or oral spray the cation is incorporated at the desired user concentration. In the case of a toothpaste, prophylactic paste, toothpowder, lozenge or chewing gum, it may be necessay, depending on the nature of the	-
10	ingradients in the composition, to increase the concentration of the cation to above the level of the desired user concentration, for example by up to five times the desired user concentration, in order to allow for incomplete availability of the cation in use as a result of specific binding of the cation to one or more of the ingredients. In these formulations it is preferable to use ingredients which avoid precipitation of the cation in the form of an insoluble salt.	10
15	A typical mouthwash has an aqueous base and generally incorporates a thickener end a flavour. A gel has an aqueous base and generally incorporates a gelling agent, a surfactant, a flavour and a preservative.	15
20	A toothposte hes an aqueous base and generally contains an abrasive, a binder, a thickener, a surfactant, a humectant, a flavouring agent and a sweetening agent. Specific compositions are described, by way of example only, in Examples 6 to 9. According to a further feature of the invention there is provided a composition for use in the method of cleaning teeth described above which in e form for use in a non-sequential manner.	20
25	By the words "in a form for use in a non-sequential manner" we mean, for example, that the composition is presented as a single pack which is intended to be used on its own, end is not intended to be used immediately before or immediately after a second different composition in a sequential manner. It is not, for example, part of a two-component pack, and it is not a single composition which is designed to release separately two components in a sequential manner. On	25
30	the contrary the composition of the invention is, for example, a simple aqueous solution or suspension, or e more sophisticated composition such as a mouthwash, toothpeste, prophylexis paste, toothpowder, lozenge, chewing gum or oral spray which is presented in a single pack for use on its own. The pack may optionally carry instructions on how the composition should be used for cleaning teeth according to the method of the invention, and such instructions may	30
35	indicate that the composition should be used in a non-sequential manner. The instructions will be appropriate to the neture of the composition. Thus, for example, of the composition is a toothpaste the instructions may indicate that the user should use the paste as the sole agent for brushing his or her teeth. The invention is illustrated, but not limited, by the following Examples:—	35
40	Example 1 Sixty dental students, average age 19 years, volunteered for the experiment. After having had a thorough prophylaxis, all the students were instructed to suspend oral hygiene for 3 days. To provoke plaque formation the students rinsed with 10 ml. of 15% w/v aqueous sucrose solution for 1 minute every second hour between 8 a.m. and 10 p.m. 4th eend of this period	40
45	the amount of plaque which has accumulated on the teeth was estimated by use of Plaque Index, as follows:-	45
50	No firmly attached plaque No firmly attached plaque visible, but some collected with a dental probe Slight amount of firmly attached plaque visible Extensive amount of firmly attached plaque visible.	50
55	The 15 individuals showing the highest Plaque Index values were selected for further study over a period of 4 days. At the start of this period, the participants were brought to Plaque Index = 0. During the test, no oral hygiene was allowed, and in addition to the sucrose rinses, 2 daily rinses with 10 ml. of 20 m.molar aquocus lantharum chloride solution were instituted in a group of 8 individuals, the remaining 8 acting as controls. At the end of the 4 day period, plaque estimations were made by one person, the results being statistically averaged. The whole	55
60	experiment was carried out under double blind conditions. The control group had a mean Plaque Index of 1.0 whereas the treated group had a mean Plaque Index of zero. The very small amount of deposit which was present in the treated group was very loosely attached to the tooth surface and could be blown off with en air jet. In a parallel experiment, individuals who rinsed with a 20 m.molar solution of aqueous stannous fluoride had a mean Plaque Index of 0.3.	60

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Example 2

A test panel of ten volunteers had a thorough prophylaxis to remove plaque from their teeth so that the Plaque Index measured according to Löe, J.Periodontol., 1967, 38, 610-616, was zero at the start of the experiment. The panel was provided with new toothbrushes and 5 instructed to brush their teeth each morning and evening for 30 seconds over a period of four

days, according to each individual's own habits, with a 10 ml. test solution, and then to rinse the teeth with the remainder of the test solution. The test solutions were 20 mM aqueous NaCl, 10 mM aqueous LaCl₃, 20 mM aqueous LaCl₃ and 20 mM aqueous YCl₃. The whole experiment was carried out under double blind conditions, each person using one of the test solutions in

10 turn. Each person's Plaque Index was brought to zero before using a new test solution. The results obtained were as follows:-

Volunteer	Unteer Plaque Index				
5 Number	20 mM NaCl	10 mM LaCl ₃	20 mM LaCl ₃	20 mM YC13	1
1	0.88	0.43	0.30	. 0.41	
2	1.12	0.63	0.44	0.54	20
20 3	0.68	0.48	0.31	0.30	
4	1.09	0.63	0.45	0.59	
5	1.02	0.83	0.52	0.50	25
25 6	062	0.60	048	0.60	
7	0.88	0.62	0.32	0.48	
8	0.55	0.42	0.34	0.42	30
30 9	0.90	0.66	0.46	0.60	2
10	0.84	0.88	0.62	0.60	
1 20					
35					:

Example 3

In a parallel experiment solutions of 20 mM aqueous GdCl₃ and aqueous YbCl₃ were tested in 40 groups of four people according to the protocol described in Example 2. The results obtained were comparable with those obtained for YCl₃ in Example 2.

A similar result was obtained using 20 mM La₂(SO₄)₃ though a complete solution at this concentration was not achieved.

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A test panel of 10 volunteers all of whom had intact buccal surfaces on their upper and lower teeth, first molar to first molar inclusive, were separated into two groups (A and B) having five persons in each.

On Day O each person's mouth and teeth was stained with a plaque disclosing agent, erythrosine and then each person had a thorough prophylaxis in order to reduce to zero the Gingival margin Plaque Index measured according to Harrap, J.Clin.Periodontol., 1974, 1,166-174.

On Days 1 and 2 each person was instructed to brush their teeth as normal with their own 55 toothbrush and toothpaste. They were then re-examined to ensure the presence of healthy gingiva, and their Gingival margin Plaque Index was again reduced to zero.

On Day 3, at 7 a.m., each group brushed their teeth with 20 ml. of either a 20 mM aqueous lanthanum acetate solution or water respectively with a new soft toothbrush dipped in the respective test solution. Rinsing for 1 minute was then carried out with the remainder of the test

60 solution. No after-rinses with water were permitted. On Day 4, at 1 pm., each person's teeth was stained with disclosing agent and the Gingival margin Plaque Index scored, the scoring being performed blindly.

Days 1 to 4 of the test period were repeated, groups A and B interchanging test solutions. The results obtained were as follows, the figures in the second and third column being a

65 summation of the percentage score for each of the 24 teeth.

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J	Number	acetate	Water	Difference		
10	1	140	140 470	0 . - 80		10
10	2	390	1 .	- 660		
	3	110	770			
	4	530	1610	-1080		15
15	5	1930	2130	- 200		
	6	1170	1760	- 590		
	7	240	370	- 130		
20	8	1300	1300	0		20
	9	520	490	+ 30		
	10	750	280	+ 470		
25		L	L	L		25
30 Example 5 A subject rinsed his mouth and teeth with 10 ml. of 20 mM aqueous LaCl, solution for 1 minute. The solution tasted salty but was not unpleasant—there was no metallic taste. After					30	
35	rinsing the subject spat out pieces of organic debris. His mouth and teeth felt much cleaner and in particular his teeth felt dry and clean. On a separate occasion the same subject rinsed with 10 ml, of 20 mM aqueous stannous fluoride. The solution tasted acid and metallic. The cleansing effect experienced with stannous fluoride was not as great as that with LaCl ₂ . A different subject found that rinsing with mM aqueous LaCl ₂ solution significantly reduced the amount of tooth stain caused by smoking tobacco.				35	
40	Example 6					40
45	Mouthwash Lacl ₃ .7H Sorbitol Ethyl Ald Pepperm Ponceau	₂ 0 Solution cohol 95% v/v nint Flavour	v/v unless other 0.7 30.6 7.6 0.7 0.0	74))% v/v I		4 5

% w/v 1.0 3.0 4.0

0.05

0.5

100

Gingival margin Plaque Index

20 mM lanthanum

50 Example 7

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Dental Gel LaCl₃.7H₂O Hydroxyethylmethyl Cellulose

Isopropanol Spearmint Oil

Polysorbate 80

Purified Water to

	Example 8 Toothpaste	% w/v 1.5	
5	LaCl ₃ ,7H ₂ O Alumina Sorbitol Solution Colloidal Silicon Dioxide	33.0 30.0 1.5	5
	Hydroxyethylcellulose Isopropanol Polyoxyethylene 50 Stearate	0.5 4.0 2.0	40
10		0.1 0.05 100	10
15	Example 9		15
	Pastille Base LaCl ₃ .7H ₂ O Gelatin	% w/w 0.1 35.0	
20	Glycerol Citric Acid Sodium Benzoate	25.0 2.0 0.2 0.1	20
25	Lemon Oil Menthol Water to	0.05 100	25

CLAIMS

- 1. A method of cleaning teeth by applying thereto a cation of one or more elements selected from yttrium, scandium, lanthanum, cerium, praseodymium, neodymium, promethium, samar-
- 30 ium, europium, gadolinium, terbium, dysprosium, holmium, erbium, thulium, ytterbium and lutetium.
 - A method as claimed in claim 1 wherein the cation is the lanthanum cation.
 A composition for use in the method of claim 1 or claim 2 which is in a form for use in a
 - non-sequential manner. 4. A composition as claimed in claim 3 which is in the form of a single pack for use on its 35
 - 5. A composition as claimed in claim 3 or 4 which is a mouthwash, toothpaste, toothpowder or dental gel.

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